**CW 4- Type II Ionic Nomenclature**

Name:

Period: 2 3 4

**Directions: Write the name or the formula for these type II Binary Ionic Compounds.**

1. Fe2O3 1.
2. CuBr 2.
3. CrCl3 3.
4. Hg2F2 4.
5. Ag3N 5.
6. Iron (II) sulfide 6.
7. Gold (III) phosphide 7.
8. Lead (IV) bromide 8.
9. Tin (IV) sulfide 9.
10. Cobalt (III) oxide 10.

**Directions: Write the name or the formula for these type II Tertiary Ionic Compounds.**

1. Ni(NO3)2 11.
2. V3(PO4)5 12.
3. Cu(OCN)2 13.
4. Zn(C2H3O2)2 14.
5. Hg3(AsO3)2 15.
6. Gold (I) sulfite 16.
7. Silver (I) peroxide 17.
8. Cadmium (III) chlorate 18.
9. Iron (III) chromate 19.
10. Manganese (II) carbonate 20.