**CW 3- Type I Ionic Nomenclature**

Name:

Period: 2 3 4

**Directions: Write the name or the formula for these type I Binary Ionic Compounds.**

1. LiCl 1.
2. CaBr2 2.
3. Al2O3 3.
4. Cs2S 4.
5. Mg3N2 5.
6. Sodium bromide 6.
7. Magnesium oxide 7.
8. Gallium iodide 8.
9. Potassium selenide 9.
10. Strontium phosphide 10.

**Directions: Write the name or the formula for these type I Tertiary Ionic Compounds.**

1. Ba(NO2)2 11.
2. K3PO4 12.
3. Be(OH)2 13.
4. Ca(C2H3O2)2 14.
5. Rb3AsO3 15.
6. Lithium sulfate 16.
7. Magnesium oxalate 17.
8. Ammonium sulfide 18.
9. Aluminum dichromate 19.
10. Sodium carbonate 20.